

# Russian R&D in Hydrogen Energy

2<sup>nd</sup> Implementation-Liaison Committee-Meeting of the IPHE Reisensburg Castle, Germany 01 March 2004 – 03 March 2004 In present Russian R&D in Hydrogen Energy is carried out within the Programs supported by:

- Ministry of Industry, Sciences and Technology
- Ministry of Atomic Energy
- Russian Aviation and Space Agency
- Russian Academy of Sciences

Lately private investors are facing to the Hydrogen Energy (Norilskii Nikel has announced a \$40 million program in Hydrogen Energy)

### Basic R&D directions

- Novel hydrogen production technologies
- Hydrogen purification technologies
- Hydrogen storage
- Fuel Cells
- New catalytic systems
- New proton exchange membranes
- Hydrogen combustion systems
- Infrastructure
- Safety and standards

### HYDROGEN PRODUCTION

- On-board and on-site H<sub>2</sub> and syn-gas production from hydrocarbon materials on the basis of plasma catalysis
- Plasmachemical large scale H<sub>2</sub> production from methane and acid gas in non-equilibrium SHF-dischargers
- Natural gas thermal decomposition and wood processing
- New catalytic systems and small scale fossil fuel processors
- Advanced electrolysis (SPE and alkaline)
- Advanced technologies for LH<sub>2</sub> production, storage and transportation

### **PEM-electrolyzers**



Small scale production of PEM-electrolysers of different productivity (up to 10 m³/h) was organized by RRC "Kurchatov institute" together with GUP Company MATYS. At present time electrolyzers are operating at current up to 1 A/cm² with energy consumption 3,9-4,2 kW·h/m³, platinum group metals loading is 1,6-3,5 mg/cm². Proved lifetime is 20000 hours.

The goals of proposed project – development of new generation of PEM-electrolysers for different pressures up to 30 atmospheres, with specific productivity up to 2 A/cm<sup>2</sup> with reduced energy consumption (3,7-4,0 kW\*h/m<sup>3</sup>) and price (to reach same price as for alkaline electrolyzers).

New membrane materials, modified platinum and not-platinum electrocatalysts, other construction materials, optimization of heat and mass transfer, improved design of electrolysis stack and the installation will be the subject of the R&D in the frame of the project.

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### NATURAL GAS PROCESSING FOR PRODUCTION OF HYDROGEN AND PURE CARBON MATERIALS

On the basis of two-stage pyrolysis of natural gas the technology to produce hydrogen from natural gas simultaneously with pure carbon material for broad commercial applications has been developed.

Thermal decomposition of natural gas is carried out in the heated porous matrix and according to the heterogeneous mechanism the pyrocarbon is formed in the pores of the matrix. As a result a solid carbon monolith is produced.

As a porous carbon skeleton for pyrocarbon stuffing other carbon containing materials can be used (e.g. wood waste products, generally the vegetable origin waste).

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0.01	Moisture content,%	0.52
0.43	Volatile, %	1.12
0.04	Ash, %	1.54
0.02	Sulphite, %	0.30
99.56	C,%	97.38
0.25	Н, %	0.27
33.18	Calorific value, MJ/kg	32.57
1.62	Density, g/cm <sup>3</sup>	0.743

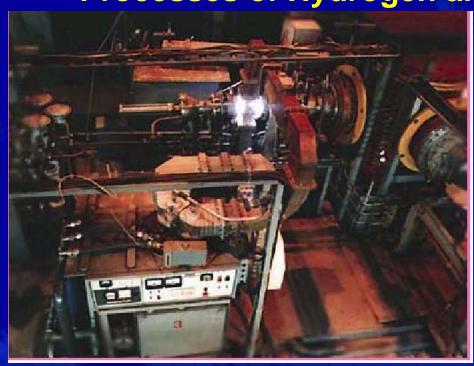


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### RRC "Kurchatov Institute's" Test Facility "POVOD" for Development and Scaling of Plasma Assisted Processes of Hydrogen and Syn-Gas Production



"FOVOD" demonstration unit at RRC "Kurchatov Institute" was designed to investigate and demonstrate the whole set of gas-phase plasma chemical processes under effect of stationary microwave discharge with power range from 10kW to 1,000 kW at microwave frequency 915 MHz. Air, nitrogen, water vapor, carbon dioxide, methane, propane-butane, oxygen, argon with a flow rate up to 2,000 m³/h at pressure between 0.005 atm – 1.0 atm can be used to burn MW discharge.

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### Methane Conversion to CO+H<sub>2</sub> in Nonequilibrium Microwave Discharge



High power microwave set-up "POVOD" includes double side symmetric microwave energy input from one hand and a high speed (150-200 m/s) gas vortex chamber with vortex number 0.7-1.2 from the other hand.

Stable discharge proved for energy input from 0.1 kWh/m<sup>3</sup> to 1.2 kWh/m<sup>3</sup>.

Experimental results have shown high efficiency methane conversion to synthesis gas ( $CO + H_2$ ) in microwave plasma both for steam and carbon dioxide reforming.

Minimum energy cost for conversion degree 50% was about 1 kWh/m³ and 1.5 kWh/m³ per 1 m³ of syn-gas. Minimum energy cost for conversion degree ~100% was less than 0.8 kWh/m³ per 1 m³ of syn-gas, in case

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### Acid Gas (H<sub>2</sub>S – CO<sub>2</sub>) Conversion in MW Discharge

Direct plasma chemical dissociation of hydrogen sulfide into hydrogen and sulfur

$$H_2S = H_2 + S$$

and syn-gas production from H<sub>2</sub>S/CO<sub>2</sub> mixture

$$H_2S + CO_2^2 > H_2 + CO + S + H_2O$$
.

have been experimentally tested under effect of different types of discharges: microwave, radio frequency, arc, gliding arc, etc.

The experimental value of energy consumption can be about 1 kWh per 1m<sup>3</sup> of H<sub>2</sub> or syn-gas, which is very close to an appropriate theoretical limit - 0.6 kWh/nm<sup>3</sup>.

The demonstration plant with discharge power up to 1 MW and productivity up to 1000 m<sup>3</sup>/h is in operation at the Natural Gas Treatment Plant in Orenburg (Russia).

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### Microwave reformers of hydrogen raw material conversion into synthesis gas

Two converter types have been developed: based on the impulse periodical microwave discharge, and on the stationary discharge.



#### Stationary converter

Synthesis gas production: 20 nm<sup>3</sup>/hour;

Converter size: below 350 x 500 x 500 mm,

Mass: 31.5 kg, converter volume: 0.1 m<sup>3</sup>.

Typical gas composition at reactor exit:

 $N_2 < 54.2\%$ ; Syngas > 40.0%;  $CO_2 < 2.5\%$ ;  $CH_4 + C_2H_4 + C_2H_2 < 3.3\%$ 

Electric power expenses for syngas production 0.15 kWh/m<sup>3</sup>

Conversion efficiency (with heat recuperation) is 94%.

Reactor: mass – 2 kg; size 80 x 400 mm Internal heat exchanger: mass – 1.5 kg; size 150 x 350 mm Microwave generator: exit peak capacity – up to 6 kW, average capacity - 3 kW, mass – 18 kg, size – 203 x 304 x 368 mm.



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# HYDROGEN SEPARATION, PURIFICATION AND STORAGE

- Systems with polymer membranes
- High-Performance Membrane Diffusion Hydrogen Purifiers
- MeHy-systems for hydrogen storage and purification
- Nano-structures for hydrogen accumulation
- Cryogenic tanks

### Surface modification of polymer membranes for gas separation



#### MAIN FEATURES:

- Method improves the selectivity of any membrane containing polydimethylsiloxane (PDMS) in the surface layers.
- Productivity of the available apparatus (5 kW) is 0.3 m per minute.
- Higher efficiency ADVANTAGES:
- prescribed membrane selectivity is available;
- 100 times improvement of selectivity;
- low treatment cost.

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### High-Performance Membrane Diffusion Hydrogen Purifiers

The Prototype modules have undergone extensive trials in real industrial environment for uninterrupted period of operation of more than 2 years without any deterioration in performance and other characteristics, without any noticeable changes in appearance, and without any loss of hermeticity.

#### **Technical Characteristics:**

Hydrogen content in feed gas

Output Hydrogen Purity

Temperature range

Tolerance to fouling feed gas impurities

Input feedstock pressure

Output pure H<sub>2</sub> pressure

Pure Hydrogen specific permeability

Membrane element diameter

Hydrogen production rate

Module weight at max production

Pure Hydrogen output per kg of module weight

30 - 98%

99.9999%

 $300 - 800 \, ^{\circ}\text{C}$ 

up to 1.5 % H<sub>2</sub>S, 15 % CO, 30 % CO<sub>2</sub>

0.15 - 10 MPa

0.1 - 3 MPa

 $20 - 300 \text{ Nm}^3/\text{m}^2 \text{ hr MPa}^{1/2}$ 

50 mm

 $1.0 - 3.0 \text{ Nm}^3/\text{hr}$   $70 - 120 \text{ Nm}^3/\text{hr}$ 

150 mm

20 kg 200 kg

 $0.1 - 1.5 \text{ m}^3$   $0.35 - 6.0 \text{ m}^3$ 

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### Metal-hydride Storage and Purification Systems





Institute for High Temperatures (IVTAN) in cooperation with Moscow Power Engineering Institute (MPEI) and Moscow State University (MSU) created experimental MeHy systems for hydrogen storage (1.5 kg H<sub>2</sub>) and purification (up to 20 nm<sup>3</sup>/h).

To optimization of design and operation of devices the two-dimensional mathematical model of heat and mass transfer in MeHy reactors has been developed.

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# Hydrogen-absorbing alloys and Nanomaterials from low-cost raw materials

New intermetallic alloys and composed nanostructures are investigated in Research Institutes of RAS and Moscow State University: MmNi<sub>5</sub>-based alloys with hydrogen content up to 1.45 mass.% and electrochemical capacity 300-330 mAh/g, and (Ti,Zr)(V,Mn,Fe,Ni)<sub>2</sub> alloys with hydrogen content up to 2 mass.% and electrochemical capacity 330-350 mAh/g. Special concern is given to the interconnection of composition and alloy structure with hydride properties, influence of redistribution of metallic components in the alloy within the homogeneity region on AB<sub>2</sub> crystal structure. The mathematical model for prediction of dependence of hydride dissociation pressure on the alloy composition was developed.

### Cryogenic hydrogen tanks characteristics

JSC "Cryogenmash"

Parameter	РЦВ- 63/0.5	РЦГ- 250/1	РЦГ- 250/0.8	PCB- 1400/1	ЦТВ- 25/0.6 t	ЦТВ- 45/1.0 t
Capacity, m <sup>3</sup>	66.3	246	246	1,437	25	45
Working pressure, MPa	0.5	1.0	0.85	1.05	0.7	1.0
Product mass, t	4.43	15.7	15.7	96.79	1.5	2.74
Evaporation loss daily, %	0.52	0.3	0.35	0.13	1.2	1.0
Dimensions, m	~9~X	7 }		V.	{ J.	3
length		36.35	36.35	3 M	14.35	15.7
width	3.68	3.7	37.4	- 1	2.75	2.5
height	12.65	3.92	3.92	20.1	3.92	3.7
Tank mass, t	22	72	72	360	19	21.76

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### **FUEL CELLS**

- •Power units on the base of Hydrogen-Air Alcaline FC
- Solid Oxide FC
- Solid Polymer FC
- •New Proton Exchange Membranes

#### **Electrochemical Generator "Cascade-IP"**



The generator is based on advanced Alkaline Fuel Cell modules. One of the unique features of the generator is a patented zero-waste regenerative scrubber used for the removal of carbon dioxide from the incoming air. The scrubber reduces the operational cost and increases the overall serviceability of the generator.

<u>Performance</u>

Maximum power: 6.0 kW; Voltage: 30.0 – 42.5 V

Maximum current: 200 A; Fuel: Hydrogen

 $H_2$  consumption < 4.5 m<sup>3</sup>/h; Air consumption < 36 m<sup>3</sup>/h

Pressure: Atmospheric

The scrubber

Operating conditions

Ambient temperature: -20 to +40 °C Relative humidity (max): 98% at 25 °C

General

Dimensions: 1250 x 582 x 863 mm; Gross weight: 240 kg

CO<sub>2</sub> scrubber: built-in, regenerative

Start-up: self-starting with built-in hydrogen

burner < 15 min at 20 °C

Electrolyte: 6.6 M KOH (aqueous solution)

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#### Planar Solid Oxide Fuel Cell





I. Solid oxide electrolyte:  $YSZ - (ZrO_2)_{0.92} (Y_2O_3)_{0.08}$ 

Diameter: 60 - 100 mmThickness:  $250 - 500 \mu\text{m}$ 

II. Electrolyte supported SOFC (operation temperature

950°C):

Electrolyte: YSZ (thickness 250 – 500 μm)

Anode: NiO + YSZ

Cathode (two layers): (I) LSM  $(La_{0.85}Sr_{0.15}MnO_3) + YSZ$ ; (II) LSM

III. Anode supported SOFC (operation temperature 800 - 850°C)

Thin-film electrolyte: YSZ (thickness  $5 - 20 \mu m$ ) Anode: NiO + YSZ (thickness  $500 - 1000 \mu m$ )

Cathorexperimentar Planar SOFC Stack (II) LSM

Number of cells: 5 - 40; Cell configuration: round or

rectangular

Operation temperature: 900 - 950°C

Stack-average power density: 200 – 250 mW/cm<sup>2</sup>

Fuel: H2 or syn-gas Oxidant: air

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# Proton exchange membranes and membrane electrode assemblies for PEMFC and DMFC

Russian research center "Kurchatov institute" together with OAO "Plastpolymer" developed perfluorinater membrane with functional sulfo-groups for PEMFC. Small scale production of such membrane was organized by extrusion technology. The parameters of this membrane are close to the Nafion ones. In recent years new methods of membrane material solution production, laboratory methods of membrane production by casting technology, new methods of the membrane electrode assemblies production were developed.

#### **Expected membrane parameters:**

Specific resistance – about 10 Ohm·cm at  $20^{\circ}$ C Thickness –  $30 - 120 \mu m$  Density –  $1.95 - 2.10 \text{ g/cm}^3$  Exchange capacity – 0.85 - 1.10 mg \* eq./g Relative elongation at rupture – 40 - 60% Water concentration 10-50%, Operating temperature – up to  $105^{\circ}$ C

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## HYDROGEN COMBUSTION SYSTEMS

### Small-scale high-pressure H<sub>2</sub>/O<sub>2</sub>-steam generators



Small-scale high-pressure  $H_2/O_2$ -steam generators are designed for use in stationary and mobile autonomous hydrogen zero emission power units (ZEPU) based on 30-150 kW mini-turbine. The main features of experimental device:

working pressure, MPa: 1-4 steam temperature, K: 600-1000 thermal capacity, kW: 40-156

length, mm: 300

max diameter, mm: 90.

<u>Fields of application:</u> transport, autonomous energy supply systems, including systems based on renewable resources (solar and wind power plants with hydrogen energy storage). In combination with steam or gas-steam mini-turbine and electric generator such systems form compact and ecologically clean power plants for various vehicles: automobiles, locomotives, ships, with efficiency well comparing with fuel cells but much cheaper.

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### High-pressure H<sub>2</sub>/O<sub>2</sub>-steam generators

Experimental high-pressure H<sub>2</sub>/O<sub>2</sub>-steam generator parameters (IVTAN and Keldysh Center):

Fuel/Oxidant SupplyGH<sub>2</sub>/LO<sub>2</sub>  $GH_2/GO_2$ Flow rate O<sub>2</sub>, g/s -1300 -1400 $H_2$ , g/s -160 -170

Steam

Temperature, °C 500-1200 500-1200 Pressure, MPa 5-7 7-10

Thermal power, MW 10-18

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### SAFETY AND STANDARDS

- Experimental investigations and mathematical modeling on Fire and Explosion Safety
- Sensors and reburners
- Safety Regulations and Standards

### Catalytic Hydrogen Reburner



#### **Features:**

Minimal hydrogen concentration in air: 0.7% (T < 70 °C),

not limited ( $T > 70^{\circ} C$ )

Water influence: relaxation time increase  $(T < 70^{\circ} C)$ 

no influence ( $T > 70^{\circ} C$ )

Relaxation time: less than 20 s ( $T < 70^{\circ}$  C)

(4vol.%  $H_2$ ): less than 1 s (T > 70° C)

Allowed humidity: up to 100%

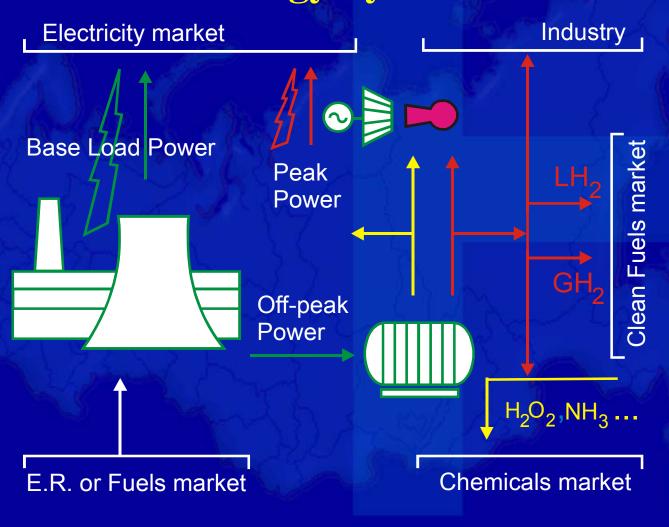
Model	Dimensions, mm	Capacity, nм <sup>3</sup> /h (4vol.%H <sub>2</sub> )
13	100×·100×·300	0.7
2	200×200×1000	7.5
3	<b>500</b> ×500×3000	100

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# INFRASTRUCTURE and SOCIAL ECONOMY STUDIES

# Integrated system of co-production of energy and chemicals from excess resources of Energy System



Liquid hydrogen Production facility. View of construction site.





Firing tests of rocket engines at tilted test stand

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